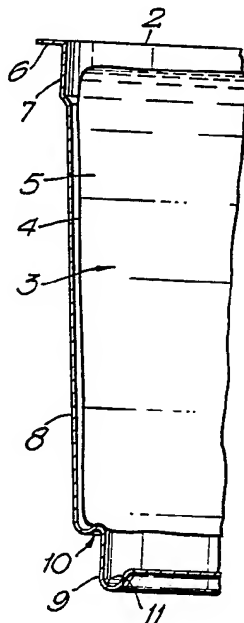




## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification <sup>4</sup> :  B65D 65/38	A1	(11) International Publication Number: WO 89/12587  (43) International Publication Date: 28 December 1989 (28.12.89)									
<p>(21) International Application Number: PCT/GB89/00669</p> <p>(22) International Filing Date: 15 June 1989 (15.06.89)</p> <p>(30) Priority data:</p> <table border="0"> <tr> <td>8814158.5</td> <td>15 June 1988 (15.06.88)</td> <td>GB</td> </tr> <tr> <td>8814159.3</td> <td>15 June 1988 (15.06.88)</td> <td>GB</td> </tr> <tr> <td>8903707.1</td> <td>17 February 1989 (17.02.89)</td> <td>GB</td> </tr> </table> <p>(71) Applicant (for all designated States except US): MAY &amp; BAKER LIMITED [GB/GB]; Dagenham, Essex RM10 7XS (GB).</p> <p>(72) Inventors; and</p> <p>(75) Inventors/Applicants (for US only): EDWARDS, David, Brian [GB/GB]; MCCARTHY, William, John [GB/GB]; ALDRED, Alan, James [GB/GB]; May &amp; Baker Limited, Dagenham, Essex RM10 7XS (GB). JACKMAN, Anthony, Douglas [GB/GB]; 14 Thorley Gardens, Pyrford, Surrey (GB).</p>		8814158.5	15 June 1988 (15.06.88)	GB	8814159.3	15 June 1988 (15.06.88)	GB	8903707.1	17 February 1989 (17.02.89)	GB	<p>(74) Agents: BENTHAM, Stephen et al.; J.A. Kemp &amp; Co., 14 South Square, Gray's Inn, London WC1R 5EU (GB).</p> <p>(81) Designated States: AT, AT (European patent), AU, BB, BE (European patent), BF (OAPI patent), BG, BJ (OAPI patent), BR, CF (OAPI patent), CG (OAPI patent), CH, CH (European patent), CM (OAPI patent), DE, DE (European patent), DK, FI, FR (European patent), GA (OAPI patent), GB, GB (European patent), HU, IT (European patent), JP, KP, KR, LK, LU, LU (European patent), MC, MG, ML (OAPI patent), MR (OAPI patent), MW, NL, NL (European patent), NO, RO, SD, SE, SE (European patent), SN (OAPI patent), SU, TD (OAPI patent), TG (OAPI patent), US.</p> <p><b>Published</b> With international search report.</p>
8814158.5	15 June 1988 (15.06.88)	GB									
8814159.3	15 June 1988 (15.06.88)	GB									
8903707.1	17 February 1989 (17.02.89)	GB									

(54) Title: PACKAGING FOR PRODUCTS



## (57) Abstract

A package for e.g. a liquid, to be dissolved or dispersed in water, contained in an envelope (3) of water soluble or dispersible material, the envelope having a flexible wall and a water soluble or water dispersible heat seal.

*FOR THE PURPOSES OF INFORMATION ONLY*

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	FI	Finland	ML	Mali
AU	Australia	FR	France	MR	Mauritania
BB	Barbados	GA	Gabon	MW	Malawi
BE	Belgium	GB	United Kingdom	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IT	Italy	RO	Romania
BJ	Benin	JP	Japan	SD	Sudan
BR	Brazil	KP	Democratic People's Republic of Korea	SE	Sweden
CF	Central African Republic	KR	Republic of Korea	SN	Senegal
CG	Congo	LI	Liechtenstein	SU	Soviet Union
CH	Switzerland	LK	Sri Lanka	TD	Chad
CM	Cameroon	LU	Luxembourg	TG	Togo
DE	Germany, Federal Republic of	MC	Monaco	US	United States of America
DK	Denmark	MG	Madagascar		
ES	Spain				

-1-

PACKAGING FOR PRODUCTS

This invention relates to a package comprising e.g. a liquid chemical or a chemical dissolved or dispersed in an organic liquid contained in an envelope of water soluble or water dispersible material and having a water soluble or water dispersible seal.

Chemicals such as pesticides and herbicides and other potentially harmful materials are often produced as a concentrated solution or dispersion in an organic liquid. Such chemicals are typically supplied in a metal or blow-moulded plastics container with a screw closure. To use the chemical pesticides or herbicides a quantity of the concentrated material is measured out of the container and then mixed with a large volume of water before being sprayed onto a locus to be treated or onto plants. Such concentrated chemical solutions are usually highly toxic so great care must be taken in measuring and mixing them to avoid splashing the liquid chemical and to avoid human or animal contact with the concentrated solution or dispersion.

Efforts have been devoted to the design of containers to minimise the risk of accidental spillage or splashing when their contents are used and also to reduce liquid residues remaining in the containers after use. Plastics containers with wide necks to facilitate pouring of their liquid contents have been used. Blow moulded plastics containers having hollow handles have been made in which the

-2-

nollow handles are isolated from the body of the container to avoid retention of liquid in the handle.

Nevertheless, with present packages it is relatively easy to spill the contents during the mixing process with the resulting risk of contamination of the environment and risk of contact with humans and animals. Also, it is rare to empty the container and farmers, and other users, tend to have partly full containers left around. These represent a further hazard. Even when all of the contents have been used it is difficult to dispose of the empty container. It is also difficult to wash adequately the containers and measuring instruments in which the concentrated solutions or dispersions are handled. These devices represent a further hazard to personnel and to the environment.

It has also been proposed to package agricultural chemicals in a container comprising a screw fitting adapted to screw onto a corresponding fitting on a spray tank. The contents of the container should be released only when a liquid-tight seal exists between the tank and the container. Practical difficulties exist in securing widespread use of such a system in view of the need for standardisation of screw fitting sizes and the possibility of leakage if a liquid-tight seal is not achieved.

It has also been proposed to package solid chemicals in water soluble containers but such containers are normally full and no particular difficulties arise in

-3-

their production or in their use.

It has also been proposed to package chemicals in a water soluble container which releases the packaged chemical only after contact with water. Such proposals have not been adopted in practice for liquid chemicals because of the limitations of known water soluble containers. Such containers have been too prone to rupture if they contained substantial quantities of liquid. It has also proved difficult to avoid pinholes at heat-sealed joints in the container and unacceptable weakness in the material of the container adjacent to heat sealed joints.

Heat sealing methods heretofore used to heat seal water soluble or water dispersible materials have not produced containers capable of retaining liquids and of withstanding the abuse to which the containers are likely to be subjected during filling, handling and storage. It has now been discovered that if the envelope material is heat sealed to yield a water soluble or water dispersible heat seal containers are obtained which possess improved abuse resistance for use in the storage of liquids.

Water insolubility or lack of water dispersibility in the material of a heat seal can arise when, for example, the heat sealing temperature is too high or the dwell time of the sealing jaws is too long. Chemical degradation of the material being sealed can then arise. Such chemical degradation which is associated with, for example, increased cross-linking of an envelope material

-4-

such as PVA reduces water solubility or water dispersibility of the material. This effect is associated with reduced abuse resistance of the containers and renders them unsuitable for liquids.

The present invention seeks to overcome the disadvantages of known packages and to provide a package which contains e.g. a non-aqueous liquid and has one or more of the following advantageous features:

The packaged chemical is released only after contact with water in which it is to be dissolved or dispersed, minimising the possibility of accidental contact of the undiluted material with the environment or with humans or animals.

The chemical can be provided in unit dosage form suitable for dilution with a predetermined amount of water removing the need for undiluted chemical to be measured out.

The packaged chemical is easy to use : the packaged chemical can be simply placed in water prior to use of the chemicals.

The need for washing out of residual chemical from containers to render them safe for disposal is removed. Containers which have been in contact with the packaged chemical remain uncontaminated which facilitates their disposal.

Substantially all of the envelope including the heat seal is water soluble or water dispersible so that no insoluble residue remains to clog or interfere with

-5-

equipment such as filters or spray nozzles.

The present invention accordingly provides a package which comprises an envelope which is, or comprises a water soluble or water dispersible material which envelope comprises a flexible wall and a water soluble or water dispersible heat seal. The package preferably contains a liquid more preferably an organic liquid.

The present invention also provides a package for a liquid which is a chemical or a solution or dispersion in an organic liquid of a chemical which liquid is contained in an envelope which is, or comprises, a water soluble or water dispersible material which is insoluble and not dispersible in the liquid and which envelope comprises a flexible wall and a water soluble or water dispersible heat seal.

It will be understood that liquid or solid chemicals can be dispersed in the organic liquid : the dispersion can be, for example, an emulsion or a suspension.

The envelope containing the chemical is preferably only partly full so that the envelope comprises an air space which generally occupies from 2 to 40% preferably from 4% to 10%, of the volume of the envelope. A larger space could be used but is less likely to be commercially attractive. Partial filling of the envelope reduces the risk of rupture of the envelope if it is subjected to shock and reduces the risk of rupture or leakage in the event of an increase in temperature which

-6-

might make the bag swell or sweat.

The volume of chemical is preferably from a half litre to two litres : a half litre is especially preferred.

Such packaging avoids the above difficulties of the prior art. To use the package an appropriate quantity of water is measured out into a vessel such as a sprayer tank and then the envelope, e.g. a bag or sachet, is removed, e.g. tipped, from the container and placed whole into the vessel with a predetermined measure of water and mixed. The contents of the envelope are released when, for example, the material from which e.g. a bag or sachet is made dissolves or disperses throughout the water together with the chemical. Thus, there is no possibility of spilling e.g. the chemical liquid since it is still in the form of a closed and sealed package when it is mixed with the large volume of water. During mixing any splashing that occurs is splashing only of a diluted chemical and this is naturally not so toxic to personnel or so damaging to the environment should any splashing or spillage occur.

The chemicals which may be packaged include those which are potentially toxic or damaging or detrimental to health or to the environment. They include pesticides for example fungicides, insecticides or herbicides (for example hydroxybenzonitrile herbicides, e.g. bromoxynil or ioxynil or derivatives thereof such as the salts or esters, e.g. heptanoates or octanoates) and, more generally, chemicals



-7-

which are to be dissolved or dispersed in a large volume of water or aqueous liquid, such as compounds, e.g. metronidazole, used to combat spoilage in industrial aqueous liquids, or compounds for addition to the aqueous circuits of e.g. domestic or industrial heating systems, compounds for addition to swimming pools, photographic materials, inks, dyestuffs, non-aqueous organic acids and cement additives. The pesticides include, e.g. molluscicides for addition to, for example, ponds or streams. When the envelope material is a PVA borates, chlorides and chlorates should generally not be present in the packaged liquid in amounts effective to lead to deterioration of the envelope material or that material should be protected from them.

Suitable water soluble or dispersible materials which are insoluble in the organic solvents used to dissolve or disperse the chemical include polyethylene oxide or methyl cellulose, but preferably the envelope, e.g. a bag or sachet, comprises or is made from polyvinyl alcohol film, i.e. partially or fully alcoholysed or hydrolysed e.g. 40-99%, preferably 70-92% alcoholysed or hydrolysed, polyvinyl acetate film.

The polyvinyl alcohol film may be unoriented, mono-axially oriented or bi-axially oriented. Water soluble materials are preferred. The materials used will generally be cold water soluble; cold water soluble PVA is preferred. It will be understood that other materials may be used when the packaged liquid is to be dissolved or dispersed in warm

-8-

or hot water and the heat seal itself can then be soluble or dispersible in warm or hot water.

The maximum tensile strength of the material of the envelope is preferably at least 20, more preferably from 30 to 80 N/mm<sup>2</sup> and the elongation at break is preferably 200 to 380%, more preferably from 220 to 350%. Testing for these values is generally carried out at 23°C and 50% relative humidity. The thickness of the envelope material is preferably from 10 to 500, more preferably 20 to 100 micrometres. Combinations of these physical properties are especially preferred.

The polyvinyl alcohol material may be extruded as a tube and then inflated to bi-axially orient it or, more preferably, it may be cast. When a cast film is used as is preferable, a tube is formed from the film and the edges heat sealed along the length of the tube. The tube is sealed at one end and then filled with the desired quantity of the chemical. The tube is again sealed above the quantity of chemical to close the envelope and form, for example, a closed bag or sachet. An air space is preferably left above the liquid in the closed envelope and, in addition, the combined volume of the air space and liquid is preferably less than the maximum possible capacity of the envelope so that it is loosely filled and can flex.

The invention also provides a process for the preparation of a package according to the invention which comprises heat sealing the envelope material to obtain a

-9-

water dispersible or, preferably, a water soluble heat seal.

When heat seals are made in order to form or close the envelope containing e.g. liquid in the package according to the invention the sealing temperature is generally from 140 to 220°C, preferably 160 to 180°C. The jaw pressure is generally from 1 to  $3\frac{1}{2}$  kg/cm<sup>2</sup>, preferably  $1\frac{1}{2}$  to  $2\frac{1}{2}$  kg/cm<sup>2</sup>. The dwell time is generally 200msec to 1.5 sec, preferably 450 msec to 1 sec.

In order to ensure optimum processability the heat sealing is generally carried out at 15 to 25°C and 15 to 85% relative humidity (RH). The relative humidity is preferably 35 to 55%. Some routine experimentation may be required to obtain suitable heat seals depending on the envelope material, e.g. the particular grade and the thickness of PVA chosen. Water solubility or water dispersibility of the heat seal can be checked by direct testing for solubility or dispersibility. The quality of the seals can also be checked by visual inspection for areas of opacity or for bubbles or, for example, by inflation of bags without liquid contents. Imperfections in the seal may give rise to a lack of water solubility or water dispersibility of the seal. The heat sealing process can be carried out on conventional heat sealing equipment which permits control and variation of the sealing jaw temperature, jaw pressure and dwell time.

In practice the envelopes according to the invention should release their contents in less than about

-10-

10 minutes. When a phytosanitary chemical is packaged the packaged chemical will be placed in the spray tank of a conventional sprayer. The tank will generally be partly filled with water, and the packaged chemical added. When the tank is provided with means to agitate the water the contents of the bag will be released more rapidly. It is preferred that release should take place in less than about a minute, for example in 30 to 40 seconds. It will be understood that the time taken to release the chemical will depend on a number of factors apart from the nature of the bag, including the temperature of the water and the level of agitation.

When the envelope is a bag or sachet the thickness of the walls should be kept to a minimum, provided that the walls have adequate strength in order to facilitate rapid dissolution or dispersion in water. A thickness of e.g. about 30 microns is particularly suitable, although large sachets may require thicker walls. The thicker the wall, the longer dissolution or dispersion of the wall material will take. It will be understood that the envelope according to the invention may comprise an area of wall which is more readily dissolved or dispersed than the rest to facilitate more rapid release of the contents of the envelope.

Suitable organic liquid solvents include petroleum based solvents, e.g. petroleum ethers, mineral oils, aliphatic or aromatic hydrocarbons, e.g. hexane,

-11-

octane, cyclohexanone, benzene, xylene and naphthalene, halogenated aliphatic or aromatic hydrocarbons, e.g. carbon tetrachloride, chloroform, methylene chloride and chlorobenzene, esters e.g. amyl acetate, ketones, e.g. cyclohexane, ethers, or a higher alcohol (lower alcohols may migrate through the water soluble or water dispersible materials described above: this can result in product appearing on the outside of the envelope). It will be understood that mixtures of solvents, e.g. mixtures of a hydrocarbon solvent with another solvent, e.g. a ketone or a higher alcohol, may also be used. The organic liquid must be reasonably dry and typically contains less than 2 to 3% of water to ensure that it does not leak prematurely from the envelope.

The liquid contents of the envelope may be thickened or rendered thixotropic. An increased viscosity in the contents can reduce the likelihood of the envelope being ruptured if the package is subjected to mechanical shock, particularly as the envelope comprises a flexible wall. The contents of the envelope may be rendered more viscous or thixotropic by the inclusion of additives, for example, a modified organophilic, or bentonite, lecithin, polymethylene oxide or silica gel.

The concentrations of pesticide or herbicide dissolved or dispersed in the organic liquid will generally be those conventionally used: in order to reduce the bulk of each envelope, however, concentrations may be increased.

-12-

Each envelope will preferably contain at least about 500 ml and will preferably contain a convenient standard volume, for example 500 ml or 1 litre, although it will be appreciated that any convenient standard volume may be chosen. The envelope will generally contain from a quarter litre to three litres of liquid although, in particular, smaller volumes may also be packaged.

Preferably the filled envelope is packaged in an outer waterproof container, for example as described in more detail in our copending application entitled "Packaging for Liquids", (which relates to a package for a liquid comprising an outer container having a shock-absorbing base and an inner water-soluble or water-dispersible envelope containing the liquid) which both protects the envelope from water and premature dissolution and also acts as a second barrier between the concentrated and potentially toxic liquid and personnel handling the container and the environment. The outer container may have the form of a container formed of plastics material with a reclosable and resealable lid containing two or more of the envelopes. Preferably however each envelope is individually packed in a separate outer container. In this case preferably the outer container is formed of thermoplastics material which is injection moulded or blow-moulded to form a container having a top, substantially flat flange, a side wall and a base. The filled envelope is placed inside the container and then a foil lid is sealed onto the top of the substantially flat

-13-

flange to provide a completely closed and sealed outer container. The lid is typically made of aluminium foil and heat sealed onto the top flange of the container but it may also be made of a plastics foil or a laminate of paper, plastics and/or aluminium.

The lid is preferably sealed to the top of the container to provide a good barrier to leakage if the envelope breaks; it is preferably bigger than the top of the container to provide a flap which can be gripped easily to remove the lid.

A laminated lid is preferred, for example a paper/aluminium/plastics laminate in which the plastics layer can be heat sealed to the lid to provide a hermetic seal. The aluminium layer provides a barrier against any pinholes which may occur in the plastics layer. The paper provides strength, processability and a label can be printed or stuck onto it. The plastics material is preferably polyethylene terephthalate (PET) which provides a good barrier against possible leakage, has good heat sealing characteristics, allows for easy removal of the lid prior to use, does not contain halogen which is potentially damaging to the environment when the lid is to be disposed of, and withstands shock. Other plastics materials can also be used e.g. polyvinylidene chloride (PVDC), polyvinyl alcohol, polypropylene or nylon.

Preferably the outside of the container is printed with information concerning the contents of the

-14-

envelope, instructions for use, and any warnings concerning the nature and toxicity of the chemical. This information may be carried on the foil lid or on a label attached to the side wall of the outer container.

The space between the envelope and the outer container (which is preferably at least about 5% of the volume of the container; the space will preferably not be more than about 30% : larger spaces could be used but may be less attractive commercially : 20 to 25% is preferred and about 25% is especially preferred. The space is preferably isolated from the atmosphere for example by a hermetic seal on the outer container. The relative humidity in the space is preferably from 45 to 70% (more preferably 50 to 60% about 50% being most preferred) at a temperature of 20°C.

When the envelope material is PVA film the mechanical properties of the film are affected by its moisture content: moisture in the film is in equilibrium with moisture both in any air space inside the envelope and in any space between the envelope and the outer container. The point of equilibrium changes with temperature so that the film may either absorb moisture or release it during storage. A relative humidity of 45 to 70% at 20°C has been found to preserve optimum storage properties of the envelope material.

The packaging in accordance with the preferred aspects of this invention provides a tough, two stage



-15-

packaging which provides for the safe transport of concentrated chemicals and allows handling of potentially toxic chemicals with the minimum risk to personnel and the environment.

The following Example illustrates the production of a package according to the invention having a water soluble heat seal:

EXAMPLE

PVA film was used to form bags containing a liquid herbicide by the following procedure using conventional bag-making equipment.

The PVA film used was SYNTANA Type KA cold water soluble PVA film, thickness 40 micrometres, with a degree of saponification of 88 mol %.

The liquid herbicide was a mixture of bromoxynil and ioxynil esters in solution in a naphthalene solvent. The liquid contained less than 3% water.

An open-topped bag was produced from the PVA film by forming the film around a shoulder and then heat sealing simultaneously the bottom and side of the bag. A jaw pressure of  $2 \text{ kg/cm}^2$  was used, with a jaw temperature of  $160^\circ\text{C}$  and a dwell time of 1 second. The ambient temperature was  $18^\circ\text{C}$  and the relative humidity 35%.

500ml of liquid herbicide was then dispensed into the bag the top of which was then sealed leaving an air space of 4 to 5% volume within the bag. Each bag was 120 mm by 205 mm and 10 bags per minute were produced.

Each filled bag was heat sealed at the top after dispensing of the liquid leaving an air space of 4 to 5% of the bag volume, the bag being about 80% full of liquid. The bag is therefore both incompletely filled and has an air space above the liquid.

Each bag was then placed into a container as

-16-

illustrated in the accompanying drawings. The container material was polypropylene. Each container was sealed using a laminated top comprising PET (polyethylene terephthalate), aluminium and paper layers. The PET layer was heat sealed to the top flange of the container leaving an air space between the bag and the container. The relative humidity in the air space was 50% at 20°C.

A further example of a package in accordance with this invention will now be described by reference to the accompanying drawings, in which:-

Figure 1 is a side elevation of the complete package;

Figure 2 is an underneath view of the outer container; and

Figure 3 shows half of a longitudinal radial section through the complete package.

The package comprises an outer container 1 having a foil lid 2 surrounding and enclosing an envelope (a bag or sachet) 3. The bag or sachet 3 is made of cold water soluble grade oriented polyvinyl alcohol film 4 made from 88% alcoholysed polyvinyl acetate having a wall thickness of 30 microns which is heat-sealed into the form of a sachet containing 500 millilitres of a concentrated dispersion 5 of a chemical in an organic liquid. The sachet 3 is housed inside the container 1 which includes a substantially flat top flange 6 connected by upper collar portion 7 to a tapering side wall 8. The container 1 also includes a foot

-17-

9 which is joined to the lowest end of the side wall 8 by a shock absorbing section 10. The container has an approximately rectangular cross-section with rounded corners between adjacent sides and with outwardly bowed faces as shown most clearly in Figure 2. The rectangular shape of the container allows relatively efficient packing together of a number of containers. The container is also tapered downwardly as shown in Fig. 1, in this case by 3 to 4° from vertical, and this allows a number of containers to be nested together when empty for easy storage or disposal. The taper also means that the sachet is supported by the walls of the container against downward movement. The container 1 is injection moulded from a block polymer polypropylene having a high melt flow index and typically having a constant wall thickness throughout of, for example, substantially one millimetre. The container is made to be translucent so that, as described below, leakage of the sachet can be detected without opening the container. In an alternative embodiment the foot only can be made translucent so that leakage can be seen. Further, polypropylene is water repellent and this makes washing of the container easier. The material from which the container is made, in this embodiment polypropylene, is sufficiently rigid to support and protect the sachet but also has a degree of flexibility which helps absorb shocks or blows to the package.

The shock absorbing section 10 is corrugated

-18-

being S-shaped in cross-section as shown in Figure 3 the ratio of the length of the section to its thickness being about 9:1, this ratio being chosen to allow the desired amount of flexing having regard to the flexibility of the material from which it is formed.

The corrugated section forms generally transverse connection joining the lower edge of the side wall 8 to the upper edge of the foot 9 which is capable of flexing as a result of the natural resilience of the thermo-plastics material to allow some relative upwards and downwards movement to occur between the foot 9 and the side wall 8. This flexing absorbs shock loads applied to the container 1 for instance if it is inadvertently dropped, e.g. during transport or handling. During transport of the packages any outer packaging containing an array of such packages may be dropped or at least suffer substantial shock loads as it is transported by, for example, lorry, or even when it is lifted and lowered by, for example, a fork lift truck. The shock absorbing portion formed between the foot and the side wall of the container flexes and absorbs such shock loads and this partly cushions the loads applied to the envelope and ensures that the outer container does not split under the application of such a shock load. Equally, after the package has been removed from any outer container the shock absorber absorbs loads if the package is, inadvertently, dropped onto a solid floor just before it is opened to gain access to the envelope. Typically, if the package falls on

-19-

its foot the shock absorbing section absorbs any shock load so imposed on the container. Equally, if the container falls on its side wall the rounded nature of the side wall means that the side wall can flex and again absorb any shock loads. Further, if the container lands upon its top flange this tends also to flex to absorb the shock load.

As can be seen from Figure 2 the foot 9 is formed with a raised central section surrounded by trough 11. The trough is provided to collect any liquid which accidentally leaks from the sachet before use. As the container, or at least this part of it, is translucent it is possible for the user to look at the foot and see whether the sachet has leaked before opening the container. Thus accidental contact with leaked contents can be avoided. As will be appreciated from Fig. 2 the raised control portion of the foot leaves a space underneath it which, when the container is stood on a shelf, would form a closed cavity. This could cause problems because if the sachet does leak into the container then vapour from the concentrate could pass through the container material into the cavity where it would be trapped and could attack the shelf or any coating on the shelf. Thus, to allow ventilation of this cavity the underside of the trough 11 is formed with at least one recess or groove (not shown) directed radially of the foot.

In this embodiment the corrugated section 10 also provides an internal annular shoulder to the container upon which the sachet rests. The curve of the corrugation

-20-

provides a smooth surface which will not strain or puncture the sachet. The sachet is thus supported above the bottom of the container resulting in further isolation from mechanical shock. The sachet can also flex under stress into the space to absorb shock.

The interior of the container is deliberately made to be smooth so as to allow the sachet to slide easily out of the container for use.

The sachet 3 is held inside the container 1 by the foil lid 2 which is heat-sealed onto the flange 6 of the container 1 or which may alternatively be connected by an adhesive.

The foil lid 2 in this embodiment is made from a heat sealable laminated material such as a polyethylene terephthalate/aluminium/paper laminate and is larger than the external diameter of the flange 6 to leave a large flap around the container which can be used to tear-off the lid.

The outer container 1 and lid 2 provide protection for the sachet 3 and so protect it from contact with water and hence its premature dissolution. It also provides an additional barrier layer around the concentrate 5 inside the bag or sachet 3 to provide an additional barrier in case of rupture of the bag or sachet 3 which prevents the potentially harmful chemical 5 from contact with personnel or the environment. However, to use the concentrate, the foil lid 2 is simply removed and then the sachet, still sealed, is dropped into a sprayer tank

-21-

containing a predetermined amount of water. The material 4 of the bag or sachet dissolves rapidly in the water so allowing the contents 5 to be dispersed throughout the water in the sprayer tank on mixing. The outer container 1 is not contaminated with the concentrated chemical and can thus be disposed of without taking any special precautions and the personnel dealing with the concentrated chemical never come into contact with it, so reducing the hazards and risks involved in handling such potentially harmful materials.

-22-

CLAIMS

1. A package which comprises an envelope which is, or comprises, a water soluble or water dispersible material which envelope comprises a flexible wall and a water soluble or water dispersible heat seal.
2. A package according to claim 1 which contains a liquid.
3. A package according to claim 2 which contains an organic liquid.
4. A package according to claim 1, 2 or 3 for a liquid which is a chemical or a solution or dispersion in an organic liquid of a chemical which liquid is contained in an envelope which is, or comprises, a water soluble or water dispersible material which is insoluble and not dispersible in the liquid and which envelope comprises a flexible wall and a water soluble or water dispersible heat seal.
5. A package according to any one of claims 1 to 4 in which the heat seal is water soluble.
6. A package according to any one of claims 1 to 5 in which the envelope comprises an air space.
7. A package according to claim 6 wherein the air space is 4 to 10% of the volume of the envelope.
8. A package according to any one of the preceding claims which comprises from a half litre to two litres of



-23-

liquid.

9. A package according to any one of the preceding claims wherein the maximum tensile strength of the material of the envelope is at least  $20 \text{ N/mm}^2$  and the elongation at break is from 200 to 380%.

10. A package according to any one of claims 1 to 8 wherein the maximum tensile strength of the envelope material is from 30 to  $80 \text{ N/mm}^2$  and the elongation at break is from 220 to 350%.

11. A package according to any one of the preceding claims wherein the thickness of the envelope material is from 10 to 500 micrometres.

12. A package according to any one of claims 1 to 10 wherein the thickness of the envelope material is from 20 to 100 micrometres.

13. A package according to any one of the preceding claims in which the material of the envelope comprises polyethylene oxide or methyl cellulose.

14. A package according to any one of claims 1 to 13 in which the material of the envelope comprises polyvinyl alcohol.

15. A package according to any one of the preceding claims in which the contents are released in less than about 10 minutes after contact with the water.

16. A package according to any one of the preceding claims in which the contents are released in less than about 1 minute.

-24-

17. A package according to any one of the preceding claims in which the organic liquid comprises a petroleum based solvent, a mineral oil, an aliphatic or aromatic hydrocarbon, a halogenated aliphatic or aromatic hydrocarbon, an ester, a ketone, an ether or a higher alcohol.

18. A package according to any one of the preceding claims in which the liquid contents of the envelope comprise additive to increase the viscosity of the liquid.

19. A package according to claim 18 in which the liquid contents comprise a modified organophilic, or bentonite, lecithin, polymethylene oxide or silica gel to increase the viscosity.

20. A package according to any one of the preceding claims which comprises, as the chemical, a compound which is potentially toxic or damaging or detrimental to health or the environment.

21. A package according to any one of the preceding claims which comprises a pesticide.

22. A package according to any one of the preceding claims which comprises a fungicide, insecticide or herbicide.

23. A package according to any one of the preceding claims in which the chemical comprises a hydroxybenzonitrile herbicide.

24. A package according to claim 23, in which the hydroxybenzonitrile herbicide comprises a mixture of ioxynil

-25-

and bromoxynil esters.

25. A package according to any one of claims 1 to 20, which comprises, as the chemical, a compound to combat spoilage in industrial aqueous liquids, a compound for addition to the aqueous circuit of an industrial or domestic heating system, a compound for addition to a swimming pool, a photographic chemical, an ink or dyestuff, a non-aqueous organic acid or a cement additive.

26. A package according to claim 1 substantially as hereinbefore described.

27. A process for the preparation of a package according to claim 1 which comprises heat sealing the envelope material to obtain a water soluble or water dispersible heat seal.

28. A process according to claim 27 in which the envelope material comprises polyvinyl alcohol.

29. A process according to claim 27 or 28 in which the material is heat sealed to obtain a water soluble heat seal.

30. A process according to claim 27, 28 or 29 which is carried out in an atmosphere having a relative humidity of 15 to 85%.

31. A process according to claim 30 in which the relative humidity is from 35 to 55%.

32. A process according to any one of claims 27 to 31 in which the sealing temperature is from 140 to 220°C.

-26-

33. A process according to any one of claims 27 to 32 in which the dwell time is from 200 msec to 1.5 sec.

34. A process according to any one of claims 27 to 33 in which the sealing pressure is from 1 to 3.5 kg/cm<sup>2</sup>

35. A process according to claim 27 substantially as hereinbefore described.

1/1

Fig.1.

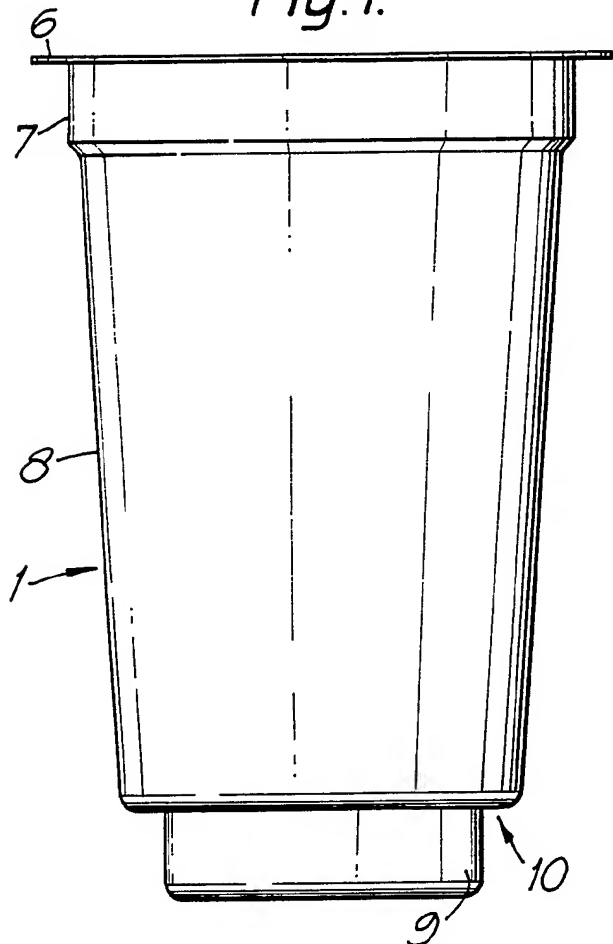


Fig.3.

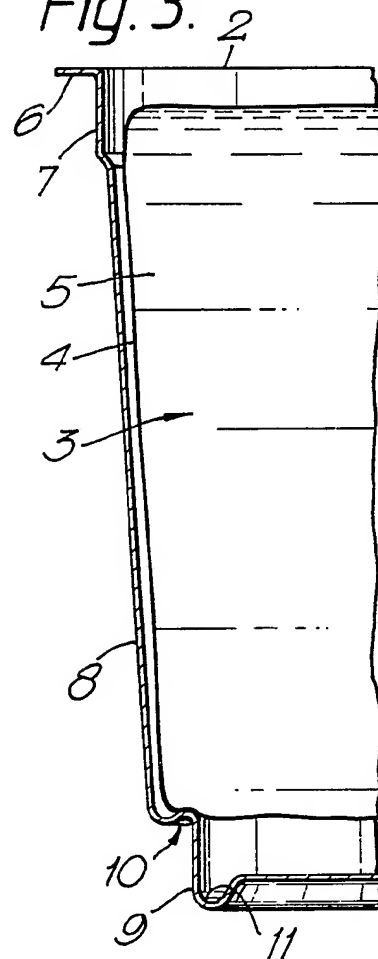
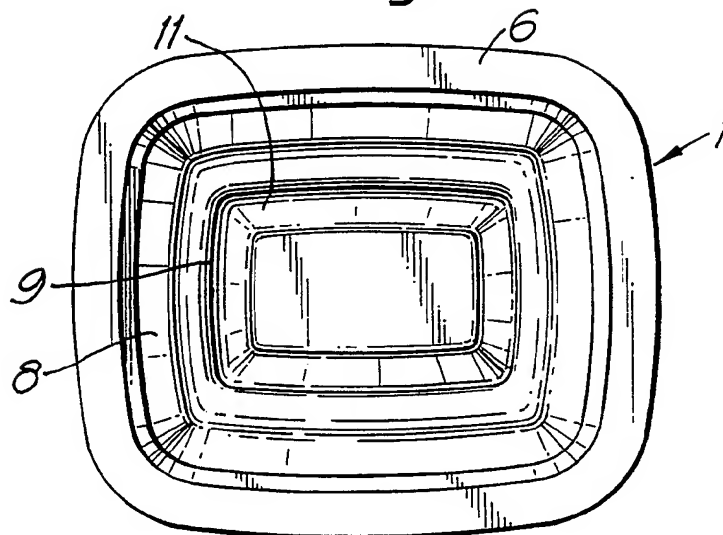


Fig.2.



# INTERNATIONAL SEARCH REPORT

International Application No PCT/GB 89/00659

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (If several classification symbols apply, indicate all) * According to International Patent Classification (IPC) or to both National Classification and IPC IPC 4 B 65 D 65/38		
<b>II. FIELDS SEARCHED</b> Minimum Documentation Searched? _____ Classification System   Classification Symbols IPC 4   B 65 D, A 01 N, B 65 B		
Documentation Searched other than Minimum Documentation to the extent that such Documents are included in the Fields Searched * _____		
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT</b> *		
Category *	Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup>	Relevant to Claim No. <sup>13</sup>
X	GB, A, 922317 (ASSOCIATED FUMIGATORS) 27 March 1963 see page 1, lines 8-24, 32-36, 43-51; page 2, lines 3-15	1-5, 8, 11, 12, 14, 17, 20-24, 26- 29, 35
Y	--	6, 9, 10, 13, 15, 16, 18, 19, 25, 30, 32, 33
Y	EP, A, 0132726 (HENKEL) 13 February 1985 see abstract, page 4, lines 19, 20	6
A	--	7
Y	Finch: "Polyvinyl Alcohol, Properties and Applications", 1973, Wiley, (London, GB), pages 362-369 see page 369, table 14.3	9, 10
Y	LU, A, 62159 (GILLET) 9 November 1971 see page 1, paragraph 1 - page 2, paragraph 4; page 4, paragraph 2	13
* Special categories of cited documents: <sup>10</sup> "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. "Δ" document member of the same patent family		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search 11th September 1989		Date of Mailing of this International Search Report 11 OCT. 1989
International Searching Authority EUROPEAN PATENT OFFICE		Signature of Authorized Officer <b>T.K. WILLIS</b>

III. DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)		
Category *	Citation of Document, with indication, where appropriate, of the relevant passages	Relevant to Claim No
Y	Finch: "Polyvinyl Alcohol, Properties and Applications, 1973, Wiley, (London, GB), pages 381-384 see page 383, paragraph 14.6.4.3; page 384, table 14.18 --	15,16,32
Y	US, A, 3030266 (CUILLE + GUYOT) 17 April 1962 see column 3, line 71 - column 4, line 5 --	18,19
Y	EP, A, 0244084 (BALL) 4 November 1987 see abstract; column 7, lines 42-52 --	25
Y	World Patent Index, Derwent Access no. 78-30769A/17, & JP, A, 53026868 (SUMITOMO CHEMICAL K.K.) 13 March 1978 --	30
A	--	31
Y	Modern Packaging, volume 58, no. 8, April 1965, McGraw-Hill, (New York, US), R.J. Ridgway et al.: "Heat sealing of polyolefins", pages 198-200 see page 143, table VIII	33
A	-----	32

**ANNEX TO THE INTERNATIONAL SEARCH REPORT  
ON INTERNATIONAL PATENT APPLICATION NO.**

GB 8900669  
SA 29507

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on 04/10/89  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
GB-A- 922317		None	
EP-A- 0132726	13-02-85	DE-A- 3326249 DE-A- 3474432	31-01-85 10-11-88
LU-A- 62159	09-11-71	None	
US-A- 3030266		None	
EP-A- 0244084	04-11-87	None	